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7-2 Review and Reinforcement. Covalent Bonding. If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided. ____F, covalent_ 1. A group of atoms united by ionic bonds is called a molecule. ____T____ 2. A covalent bond is formed by a shared pair of ...

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7-2 Review and Reinforcement (continued) Write Lewis structures for each of the following molecules. Indicated the bonds with either dots or dashes. 15. NH3 18. PCl3. 16. H2 19. CCl4. 17. C2H4 20. C2H6. Answer the following questions as directed. 21. Explain why the molecules SF4 is an exception to the octet rule.\\

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6—2 Review and Reinforcement (continued) Identify each of the following elements as a meta/ (M), nonmetal (NM), or semimetal (SM). 19. sodium —SM— silicon 21. neon 22. 23. nitrogen Write the family names that have been given to each of the following groups. 24. Group IA alkan- 25. Group 2A 26. Group 7 A 27. Group8A hvohit

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